掺铬锂锰氧化物的湿化学法合成与性能

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摘要:为了寻求新的合成方法,改善锂锰氧化物的电化学性能,采用湿化学法合成了掺铬尖晶石锂锰氧化物 Li Mn2.xCrxQ4 (x=0.00,0.04,0.08,0.12),并通过热重 - 差热(TGDTA)、X射线衍射(XRD)、扫描电镜(SEM)与恒电流充放电等测试 方法考查了材料的结构 形貌与电化学性能。 TG DTA 测试证明:湿化学法所制得的 Li $\mathrm{Mn}_{2-x}\mathrm{Cr}_x \mathbb{Q}_4(\ x=0.04)$ 前驱体在 100~900 ℃之间有一个较大的放热峰 ,在此温度范围内 ,前驱体由非晶态向晶态转变。 XRD 结果表明 :合成产物具有尖晶 石结构,随掺铬量的增加,样品的晶格常数变小,晶胞体积收缩。SEM观测结果表明:合成材料的颗粒粒度较小,分布均 匀 ,结晶性能良好 ,掺铬后样品(x = 0. 04) 的粒径为 0. 4 μ m ,大于未掺铬时的 0. 2 μ m 。恒电流流放电实验结果表明 :随掺铬 量的增加,样品的首次放电比容量降低,当掺铬量由 x=0.00 增至 x=0.12 时,初始放电比容量由 $123.3\,$ m Ah/g 降至 113.6 m Ah/g。兼顾比容量与循环性能,以 Li Mn_{1.92} Cr_{0.08} O₄ 性能最佳。

关键词:掺铬; 尖晶石锂锰氧化物; 湿化学法; 电化学性能

文章编号:1001 - 1579(2004)04 0263 - 03 中图分类号: T M912.9 文献标识码:A

Synthesis and performance of chromium doped spinel lithium manganese oxide prepared by wet chemical method

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Abstract: In order to seek a new synthic method and improve the electrochemical property of lithium manganese oxide, chromium doped spinel lithium manganese oxides Li Mar $_x$ Cr $_x$ O₄(x = 0.00, 0.04, 0.08, 0.12) were synthesized by wet chemical method. The structures, morphologies and electrochemical performance of the products were studied by means of modern method, such as thermal gravimetric and differential thermal analysis (GDTA), X-ray diffraction (XRD), scanning electron microscope (SEM) and constant current cyclic test. There was a large exothermic peak corresponding to the transition from an amorphous structure to a crystal one within the temperature rang of $100 \sim 900$ The DTA curves of the precursor of Li Mn_{2-x}Cr_xO₄(x = 0.04). The XRD results showed that all the same ples had a spinel structure and the lattice parameters decreased with the increasing amount of the chromium doped. The SEM micrographs of the products revealed that the particle size distribution was narrow and the average grain sizes of Li Mn₂ O₄ and Li Mn₂ . $_x$ Cr $_x$ O₄ (x =0.04) were 0.2 µm and 0.4 µm, respectively. The results of constant current cyclic tests showed that the cyclic performance of the products increased with the amount of chromium doped, while their initial capacities decreased from 123.3 mAh/g of Li Mn_2 O_4 to 113.6 mAh/ g of Li Mn_{2-x}Cr_xO₄(x=0.12). According to both the special capacity and the cyclic performance, the product with chromium doped of x = 0.08 had the optimized performance.

Key words: chromium-doped; spinel lithium manganese oxide; wet che mical method; electroche mical performance

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